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# Synthesis of Terminal-Biradical Compounds Consisting of Two N-Oxyl Groups Connected with Conjugated $\pi$ -Systems

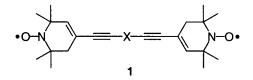
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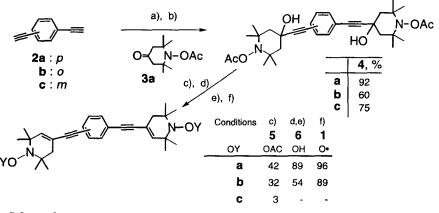
**Abstract:** Terminal-biradical compounds 1 having two N-oxyl groups connected with conjugated  $\pi$ -systems were synthesized by (1) addition of magnesium acetylide to 4-oxo-TEMPO, followed by dehydration, or (2) cross-coupling between acetylene and vinyl triflate derived from 4-oxo-TEMPO. © 1997 Elsevier Science Ltd.

The design and synthesis of molecular magnets are one of the current interests.<sup>1</sup> Essentials are how to increase the spin density and to regulate the direction of the spins. For latter purpose, the radicals are attempted to be joined by  $\pi$ -electron systems to allow to interact each other. Poly-radical compounds connecting the radicals by aromatic structure have been reported.<sup>1a,b,d</sup> However, the overlap of the aromatic  $\pi$ -orbital seems not so efficient as expected because of a steric repulsion. We planed to connect radicals by other type of  $\pi$ -system, *e.g.*, acetylenic and vinylic moieties. Herein, we describe preliminary synthetic results on terminal-biradical compounds 1 containing two *N*-oxyl groups connected with conjugated  $\pi$ -systems by (1) addition of magnesium acetylide to 4-oxo-TEMPO followed by dehydration, or (2) Pd-catalyzed cross-coupling between acetylene derivatives and vinyl triflate derived from 4-oxo-TEMPO.



Diacetylenes 2, obtained by the cross-coupling of dibromobenzenes and 3-methyl-1-butyn-3-ol in the presence of a  $PdCl_2(PPh_3)_2$ -CuI catalyst followed by removal of the 2-propanol moiety with KOH in toluene at 100 °C, underwent metallation with ethylmagnesium bromide and subsequent reaction with *N*-acetoxy-2,2,6,6-tetramethylpiperidine<sup>2</sup> 3a to give the corresponding diols 4 in 60-92% yields. The diols 4a<sup>3</sup> and 4b were converted to the terminal biradicals 1a and 1b by 1) acid-catalyzed dehydration<sup>4</sup> with *p*-TsOH, 2) base-

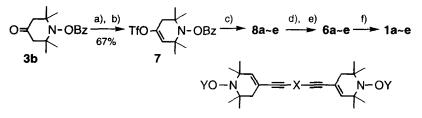
promoted hydrolysis, and 3) oxidation with lead dioxide in good to moderate yields. In contrast, the dehydration of the *meta*-diol **4c** gave the corresponding dienediyne **5c** in poor yield (Scheme 1).



## Scheme 1

a) EtMgBr (2 equiv.), THF, room temp., 1 h; b) 3a (2 equiv.), room temp., 20 h; c) *p*-TsOH (0.5 equiv.), PhH, reflux, 6 h; d) 1 M NaOH, MeOH, 50 °C ~ reflux, 6 h; e) H<sub>3</sub>O<sup>+</sup>; f) PbO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, room temp., 10 h.

Therefore, more versatile methods are now required to develop for the synthesis of the desired compounds 1. The different approaches to arrive at the targets are as follows: firstly, *N*-(benzyloxy)-2,2,6,6-tetramethylpiperidine  $3b^2$  was allowed to convert into the corresponding vinyl triflate 7 (67% yield). Actually, this triflate is stable enough to store for a few months in a refrigerator. Cross-coupling<sup>5</sup> of 7 with diacetylene compounds 2 in the presence of triethylamine and a catalytic amount of PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> in DMF gave the corresponding dienediyne 8 in good to moderate yields. Hydrolysis of 8 followed by oxidation afforded the desired biradical 1 (Scheme 2, Table 1).



8a~e (Y = Bz); 6a~e (Y = H); 1a~e (OY =  $O^{\bullet}$ )

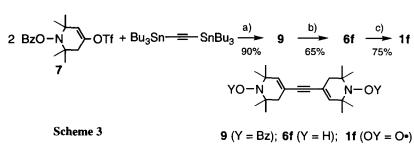
### Scheme 2

a) LDA, THF, -78 °C, 2 h; b) Tf<sub>2</sub>NPh, 0 °C, 5 h; c)  $\equiv$ -X- $\equiv$  (2, see Table 1), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, Et<sub>3</sub>N, DMF, 50 °C, 2 ~ 4.5 h; d) 1 M NaOH, MeOH, 50 ~ 80 °C, 2.5 ~ 11 h; e) H<sub>3</sub>O<sup>+</sup>; f) PbO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, room temp., 10 ~ 12 h.

	Compounds <sup>a</sup>	Isolated Yield, %			1
	Z-X-Z	<b>8</b> (Y = Bz)	6 (Y = H)	1 (OY = O•)	Mp., °C <sup>b</sup>
a	z-∕∑-z	57	44	96	220 ~ 221 (decomp.)
Ь	کے z z	54	45	89	47 ~ 48
С	z	70	43	81	147 ~ 149
d	z-√_}−z	59	72	74	164 ~ 165 (decomp.)
е	z-√s)–z	53	70	69	177 ~ 178 (decomp.)
a) $Z = YO - N$ . b) The melting points were not corrected.					

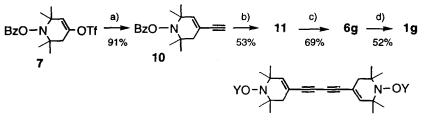
Table 1. Synthesis of Biradicals 1 *via* Cross-Coupling of 7 with 2, Hydrolysis of 8, and Oxidation of 6

The other acetylene derivatives **1f** and **1g** bearing the dehydropiperidinyl radical moiety attached to the both side of acetylene were synthesized by use of the Pd-catalyzed cross-coupling reaction. 1,2-Bis(tributylstannyl)ethyne was coupled<sup>6</sup> with the triflate **7** in the presence of Pd catalyst to afford dienyne **9** in 90% yield. Hydrolysis followed by oxidation gave the desired biradical **1f** (Scheme 3).



a) Pd(PPh<sub>3</sub>)<sub>4</sub>, LiCl , THF, 60 °C, 2.5 h; b) 1 M NaOH, MeOH, 70 °C, 5 h;
c) PbO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, room temp., 15 h.

Cross-coupling of the triflate 7 with tributylstannylethyne afforded the enyne 10 in 91% yield. Pdcatalyzed homo-coupling<sup>7</sup> of 10, followed by hydrolysis and oxidation, gave the desired biradical 1g (Scheme 4).



#### Scheme 4

11 (Y = Bz); 6g (Y = H); 1g (OY =  $O^{\bullet}$ )

a) =-SnBu<sub>3</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>, LiCl, THF, 60 °C, 2 h; b) Pd(PPh<sub>3</sub>)<sub>4</sub>, Cul, NEt<sub>3</sub>, ClCH<sub>2</sub>COCH<sub>3</sub>, PhH, room temp.; c) 1 M NaOH, MeOH, 80 °C, 8 h; d) PbO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, room temp., 14 h.

In conclusion, we have synthesized the terminal biradicals 1 by addition of magnesium arenedi(acetylide) to 4-oxo-TEMPO, followed by dehydration. We also now offer the vinyl triflate 7 as a new versatile intermediate for the preparation of N-oxyl-dehydropiperidine derivatives by cross-coupling between acetylene or stannylacetylene compounds. Spin-spin interaction between radicals in 1 was observed by electrochemical and spectroscopic measurements. The detail will be submitted elsewhere. Korshak reported the synthesis of polymer material derived from diol analog of 1g and suggested that control of the superstructure of the polymer is critical to manifest high ferromagneticity.<sup>8</sup> Further extended investigations are continued in our laboratories.

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## **References and Notes**

- a) Mitsumori, T.; Inoue, K.; Koga, N.; Iwamura, H. J. Am. Chem. Soc. 1995, 117, 2467. b) Inoue,
   K.; Iwamura, H. J. Am. Chem. Soc. 1994, 116, 3173. c) Yoshioka, N.; Lahti, P. M.; Kaneko, T.;
   Kuzumaki, Y.; Tsuchida, E.; Nishide, H. J. Org. Chem. 1994, 59, 4272. d) Kanno, F.; Inoue, K.;
   Koga, N.; Iwamura, H. J. Am. Chem. Soc. 1993, 115, 847. e) Inoue, K.; Koga, N.; Iwamura, H. J.
   Am. Chem. Soc. 1991, 113, 9803. and references cited therein.
- These TEMPO derivatives 3a and 3b are prepared from commercially available 4-oxo-2,2,6,6-tetramethylpiperidine N-oxyl through reduction of N-oxyl (PhNHNHPh/CH<sub>2</sub>Cl<sub>2</sub>, room temp., 1 h, 95%), and protection of the N-hydroxyl group with Ac<sub>2</sub>O/Py/cat. DMAP/THF (room temp., 2.5 h, 3a, 90%) or Bz<sub>2</sub>O/Py/CH<sub>2</sub>Cl<sub>2</sub> (0 °C, 5 h, 3b, 98%).
- 3. p-Disubstituted diacetate 4a was also prepared (isolated yield: 95%) from cross-coupling of 1-acetoxy-4-ethynyl-4-hydroxy-2,2,6,6-tetramethylpiperidine, derived from N-acetyloxy-4-oxo-TEMPO 3a, and p-diiodobenzene in the presence of catalytic amount of CuI and PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> in NEt<sub>3</sub> at room temperature. However, no desired 4a could be obtained from cross-coupling with p-dibromobenzene.
- 4. Mesylation (MsCl/pyridine(/CH2Cl2)) and tosylation (p-TsCl/pyridine(/CH2Cl2)) did not proceed at all.
- 5. Ritter, K. Synthesis 1993, 735.
- 6. Scott, W. J.; Crisp, G. T.; Stille, J. K. J. Am. Chem. Soc. 1984, 106, 4630.
- 7. Rossi, R.; Carpita, A.; Bigelli, C. Tetrahedron Lett. 1985, 26, 523.
- 8. Korshak, Y. V.; Medvedeva, T. V.; Ovchinnikov, A. A.; Specktor, V. N. Nature 1987, 326, 370.

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